The Synthesis of Novel Methotrexate-like Compounds

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Summary. Methotrexate (*MTX*) is a folate antagonist used in treatment of several chronic inflammatory and neoplastic conditions. In this study, new *MTX*-like compounds that maybe potential anticancer agents were synthesized and their structures were determined by IR, UV, GC-MS, ¹H NMR, and ¹³C NMR spectra. Also, in this study, a series structurally related to *MTX* or folate analogous compounds were evaluated whether they have inhibitory properties on the dihydrofolate reductase activity (DHFR).

Keywords. Methotrexate; *p*-Aminobenzoic acid; *DMAP*; *DCC*; Amino acid esters.

Introduction

Methotrexate (MTX; 2,4-diamino- N^{10} -methylpteroylglutamic acid; I) is a small molecule inhibitor of the ubiquitous enzyme dihydrofolate reductase (DHFR), which is a requisite component of the biosynthesis pathway for purine and pyrimidine. MTX is a classical type of inhibitor of dihydrofolate reductase and is known to be metabolized to polyglutamyl forms after its incorporation into cells [1]. MTX is also a potent folic acid antagonist which inhibits the reduction of folic acid to tetrahydrofolic acid, thereby leading to the suppression of the nucleic acid biosynthesis pathway, and cell death. It has been used in the treatment of neoplastic disorders since the late 1940s [2, 3]. As a consequence, MTX is used as a drug for the treatment of a wide range of diseases, e.g., for cancer treatment since its development and the identification of its biological activities [4].

High doses (>3 g/m²) have been prescribed as a cytostatic agent in oncology since 1953 [5]. Low doses of methotrexate (5-25 mg/m² per week) have been used increasingly for the treatment of patients with rheumatoid arthritis (RA) [6, 7], or psoriasis [8–12] and steroid-dependent asthma [13–15]. At the same time *MTX* is widely utilized in the treatment of acute leukemia, carcinoma of the breast, osteosarcoma, ovarian cancer, and other malignant diseases [16].

Therapeutic drug monitoring is of importance in order to improve *MTX*–RA therapy because of the severe side-effects of *MTX*, *e.g.*, marrow suppression, gastro-intestinal lesion, renal insufficiency, hepatic failure, hypoalbuminemia, and pancytopenia [17, 18].

MTX is known for its severe serious side effects although it is used in a wide range in medicine. Because of these shortcomings the synthesis of MTX-like compounds is pursued. Aim of this work is the synthesis of compounds which can be used instead of MTX and having less side effects than MTX. Therefore we use some aromatic aldehydes instead of the pterin ring, but did not alter the PABA and glutamic acid parts. Then we use β -alanin instead of glutamic acid.

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$$Ar$$
-CH=N- CO_2H + RNH_3CI $OCC/DMAP$ OCC_2H + RNH_3CI OCC_3H OC

Scheme 1

Results and Discussion

Compounds 3a-3j were synthesized as depicted in Scheme 1. The -COOH functionalized imines 1 were reacted with amino acids (β -alanine and L-glutamic acid) with DCC and DMAP as coupling agent.

The IR spectra of these compounds show the C=N bond stretching vibration in the $1625-1635 \,\mathrm{cm^{-1}}$ and the Ar-H bond stretching vibration in the $3101-3037 \,\mathrm{cm^{-1}}$ region and the strong sharp absorption bands of the amide (NH) appeared at $3307-3435 \,\mathrm{cm^{-1}}$. The 1 H NMR spectra of the compounds were recorded in DMF-d₇. The 1 H NMR assignments are detailed in the following. The imine protons for compounds 3a-3j are observed as singlets of $9.08, 9.00, 9.08, 8.60, 9.88, 9.26, 9.00, 9.08, 8.34, and <math>9.87 \,\mathrm{ppm}$.

Also this series of compounds was evaluated whether they have inhibitory properties on the dihydrofolate reductase activity (DHFR).

The result of screening regarding the effect of methotrexate related compounds on DHFR activity is shown in Table 1, listing the compounds in their rank order of potency as inhibitors of DHFR activity. As shown in Table 1, compounds **3g**, **3e**, **3j**, and **3d**, with inhibitions of 84, 83, and 82%

Table 1. Maximal inhibition of DHFR by selected compounds **3a-3j** related to methotrexate

Test substance	Maximal inhibition in % (mean)
Methotrexate	84
3a	64
3 b	72
3c	78
3d	82
3e	83
3f	70
3 g	84
3h	74
3i	78
3ј	82

(mean), were the most potent inhibitors among the compounds.

Experimental

Melting points were measured on a Gallenkamp apparatus using a capillary tube. Infrared absorption spectra were recorded from a Mattson 1000-FTIR spectrometer, using KBr pellets. ¹H NMR and ¹³C NMR spectra were obtained with a Bruker DPX FT-NMR (300 MHz) spectrometer. The chemical schifts are reported in δ/ppm with external TMS. Mass spectra were determined on a Thermo Finnigan Trace DSO. The elemental analyses were performed on a LECO CHNS-932 C, H, N analyzer. All were found to agree favourably with the calculated values. All reagents were of analytical grade. Salicylaldehyde, 2-hydroxy-4-methylbenzaldehyde, thiophene-2-carbaldehyde, pyrrole-2-carbaldehyde, 2-hydroxy-1naphtaldehyde, 4-aminobenzoic acid, glycine, β -alanine, L-glutamic acid, dichloromethane, thionyl chloride, 4-(dimethylamino)pyridin, and N,N'-dicyclohexylcarbodiimid were purchased from Aldrich Chemical Co. Ethanol, methanol, and CH₂Cl₂ were purchased from Merck (Germany). Dichloromethane was distilled from P₂O₅ before use. Amino acid esters were prepared according to published procedures [19, 20]. An improved preparation of these two esters is reported below.

β -Alanine Methyl Ester

Thionyl chloride (16.7 g, 0.14 mol) was added dropwise to a suspension of 8.91 g β -alanine (0.10 mol) in 75 cm³ methanol at 0°C. The mixture was stirred at room temperature for 18 h and then concentrated using a rotary evaporator. The residual oil occurred was triturated with cold ether and β -alanine methyl ester precipitated as a white powder. The powder was air-dried 12.38 g (89%), mp 103–104°C (Ref. [21] 103–104°C).

Dimethyl L-Glutamate Hydrochloride

Freshly distilled $6.5\,\mathrm{cm}^3$ thionyl chloride (75 mmol) were added dropwise with stirring to cold (at $0^\circ\mathrm{C}$) anhydrous $20\,\mathrm{cm}^3$ methanol. Dry L-glutamic acid (2.4 g, 14 mmol) was added in portions to this mixture and stirred for 18 h at ambient temperature. On completion of the reaction confirmed by TLC analysis (*n*-butanol:acetic acid:H₂O = 6:2:2), the reaction mixture was concentrated on a rotary evaporator to give a heavy syrup. The syrup was crystallized by the addition of dry diethyl ether to give a fine white powder. Recrystallization from acetone-diethyl ether gave $2.55\,\mathrm{g}$ (87%) dimethyl L-glutamate hydrochloride as a colorless solid, mp 77–79°C (Ref. [22] 78–80°C).

Synthesis of 4-[(2-Hydroxy-4-methoxybenzylidene)amino]-benzoic acid] (3b). Typical Procedure

A mixture of 1.52 g 2-hydroxy-4-methoxybenzaldehyde (10 mmol) in 20 cm³ *Et*OH and 1.37 g 4-aminobenzoic acid (10 mmol) in 20 cm³ *Et*OH was refluxed for 2 h and then cooled to room temperature. The solid which precipitated was collected, washed with cold ethanol, and recrystallized from ethanol to give 2.00 g (74%) of a yellow solid, mp 289–290°C.

All imine compounds were prepared according to this procedure.

Coupling of Amino Acids to Imines. General Procedure A mixture of 4-[(2-hydroxybenzylidene)amino]benzoic acid, dicyclohexylcarbodiimide (DCC) and 4-(dimethylamino)pyridine in dichloromethane was stirred at 0° C for 3 h. Then β alanine methyl ester hydrochloride was added to the mixture. After 2h, temperature of the solution was allowed to rise to room temperature and the solution was stirred at this temperature for 48 h. The precipitated N,N'-dicyclohexylurea was removed by filtration and the filtrate was extracted with water, then cold dilute HCl solution, and then 10% NaHCO₃ solution. Evaporation gave a residue which was dissolved in 2 cm³ ethanol and then hydrolyzed at room temperature with 12 cm³ cold aqueous alcoholic sodium hydroxide (0.10 M) to afford the required acid as a sodium salt. The solution of acid salt was acidified with a dilute hydrochloric acid solution (0.10 M) to obtain the product as acid. The precipitate occurred was filtrated off and crystallized from DMF-water.

The Investigation of Inhibitory Properties of 3a-3j on the Dihydrofolate Reductase (DHFR) Activity

DHFR and L-7,8-dihydrofolic acid were purchased from Fluka (from bovin liver, 9 U/cm³) and Sigma, respectively. Methotrexate and related compounds were dissolved in *DMSO* making a stock solution, which was diluted approx. 100 fold in the assay. At this concentration *DMSO* had no effect on DHFR activity as tested in control experiments. All experiments were done in duplicate with 2 parallel samples. Activity and inhibition values were estimated by the linear regression analyses Determination of DHFR activity and the inhibitor studies were carried out as described by *Pastore et al.* [23], and DHFR activities were expressed in IU/cm³. The experimental results are given below.

N-[4-[[(2-Hydroxyphenyl)methylene]amino]benzoyl]-eta-alanin (**3a**)

Yield 36%, mp 158–160°C; MS (EI): m/z = 313.27 (molecular weight 312.33); ¹H NMR (300 MHz, DMF-d₇): $\delta = 2.71$ (t, 2H, H-7′), 3.65 (q, 2H, H-6′), 7.11 (dd, 2H, H-4, H-5), 7.49 (t, 1H, H-6), 7.57 (d, 2H, H-2′, J = 8.51 Hz), 7.74 (dd, 1H, H-3), 8.09 (d, 2H, H-3′, J = 8.51 Hz), 8.65 (t, 1H, NH), 9.08 (s, 1H, CH=N), 12.62 (b, 1H, Ar–OH), 13.01 (s, 1H, COOH) ppm; ¹³C NMR (75 MHz, DMF-d₇): $\delta = 33.87$ (C-7′), 36.08 (C-6′), 112.80 (C-5), 116.83 (C-4), 117.27 (C-1), 121.41 (C-2′), 128.78 (C-3′), 133.11 (C-6), 133.75 (C-3), 150.94 (C-4′), 152.19 (C-1′), 161.11 (C-2), 165.12 (CH=N), 166.71 (C-5′), 173.20 (C-8′) ppm; IR (KBr): $\bar{\nu} = 3352$ s (CONH), 3056s (Ar–H), 2929w (C–H), 1740s (CON), 1625s (C=N) cm⁻¹.

N-[4-[[(2-Hydroxy-4-methoxyphenyl)methylene]amino]-benzoyl]-\(\beta\)-alanin (3b)

Yield 32%, mp 212–215°C; MS (EI): m/z = 343.06 (molecular weight 342.35); ¹H NMR (300 MHz, DMF-d₇): δ = 2.67 (t, 2H, H-7′), 3.68 (q, 2H, H-6′), 3.91 (s, 3H, OCH₃), 6.56 (d, 1H, H-3), 6.63 (d, 1H, H-6), 7.53 (d, 2H, H-2′, J = 8.60 Hz), 7.64 (d, 1H, H-5), 8.05 (d, 2H, H-3′, J = 8.60 Hz), 8.63 (t, 1H, NH), 9.00 (s, 1H, CH=N), 12.62 (b, 1H, Ar–OH), 13.60 (s, 1H, COOH) ppm; ¹³C NMR (75 MHz, DMF-d₇): δ = 35.78 (C-7′), 36.05 (C-6′), 55.65 (OCH₃), 107.80 (C-6), 112.80 (C-5), 113.36 (C-1), 121.16 (C-2′), 128.77 (C-3′), 132.60 (C-4′), 134.63 (C-3), 150.91 (C-1′), 152.17 (C-4), 163.83 (C-2), 163.93 (CH=N), 166.01 (C-5′), 173.18 (C-8′) ppm; IR (KBr): $\bar{\nu}$ = 3333s (CONH), 3056s (Ar–H), 2921w (C–H), 1734s (CON), 1635s (C=N) cm⁻¹.

$N-[4-(2-Thiophenealdimino)benzoyl]-\beta-alanin$ (3c)

Yield 33%, mp 168–170°C; MS (EI): m/z = 302.38 (molecular weight 302.35); ¹H NMR (300 MHz, DMF-d₇): $\delta = 2.86$ (t, 2H, H-7′), 3.85 (q, 2H, H-6′), 7.43 (dd, 1H, H-3), 7.54 (d, 1H, H-2′), 7.96 (d, 1H, H-2), 8.09 (d, 1H, H-4), 8.23 (d, 2H, H-3′), 8.79 (t, 1H, NH), 9.08 (s, 1H, CH=N), 12.73 (b, 1H, COOH) ppm; ¹³C NMR (75 MHz, DMF-d₇): $\delta = 34.08$ (C-7′), 36.24 (C-6′), 113.02 (C-3), 121.15 (C-2′), 128.73 (C-3′), 131.78 (C-4′), 132.34 (C-2), 134.37 (C-4), 143.03 (C-1′), 154.01 (C-1), 155.19 (CH=N), 166.38 (C-5′), 173.28 (C-8′) ppm; IR (KBr): $\bar{\nu} = 3435$ s (CONH), 3101s (Ar–H), 2935w (C–H), 1721s (CON), 1633s (C=N) cm⁻¹.

$N-[4-(2-Pyrrolaldimino)benzoyl]-\beta-alanin$ (3d)

Yield 34%, mp 173°C (decomposed); MS (EI): m/z = 284.80 (molecular weight 285.31); ¹H NMR (300 MHz, DMF-d₇): $\delta = 2.86$ (t, 2H, H-7′), 3.83 (q, 2H, H-6′), 6.48 (dd, 1H, H-3), 6.99 (d, 1H, H-2), 7.37 (d, 1H, H-4), 7.43 (d, 2H, H-2′, J = 8.45 Hz), 8.17 (d, 2H, H-3′, J = 8.45 Hz), 8.60 (s, 1H, CH = N), 8.73 (t, 3H, $CONHCH_2$), 12.02 (b, 1H, NH), 12.73 (b, 1H, COOH) ppm; ¹³C NMR (75 MHz, DMF-d₇): $\delta = 35.82$ (C-7′), 36.04 (C-6′), 112.87 (C-3), 121.85 (C-2′), 128.84 (C-3′), 132.19 (C-4′), 134.21 (C-4), 137.85 (C-2), 144.19 (C-1′), 153.85 (C-1), 155.03 (CH = N), 166.23 (C-5′), 173.25 (C-8′) ppm; IR (E = N) (E = N), 1676s (E = N), 1628s (E = N) cm⁻¹.

N-[4-[[(2-Hydroxy-1-naphtyl)methylene]amino]benzoyl]-eta- alanin (3e)

Yield 41%, mp 235–237°C; MS (EI): m/z = 363.53 (molecular weight 362.39); 1 H NMR (300 MHz, DMF-d₇): δ = 2.66 (t, 2H, H-7′), 3.63 (q, 2H, H-6′), 7.08 (d, 1H, H-5), 7.40 (t, 1H, H-6), 7.58 (t, 1H, H-7), 7.78 (d, 2H, H-2′, J = 7.50 Hz), 7.88 (d, 1H, H-8), 8.02 (d, 1H, H-4), 8.11 (d, 2H, H-3′, J = 7.50 Hz), 8.59 (d, 1H, H-3), 8.68 (t, 1H, CONHCH₂), 9.88 (s, 1H, CH=N), 15.72 (b, 1H, Ar–OH) ppm; 13 C NMR (75 MHz, DMF-d₇): δ = 34.04 (C-7′), 36.24 (C-6′), 109.39 (C-10), 120.59 (C-6), 120.65 (C-2′), 122.16 (C-7), 123.96 (C-5), 127.57 (C-9), 128.50 (C-8), 128.99 (C-3′), 129.40 (C-4′), 137.47 (C-3), 157.02 (CH=N), 132.68 (C-1), 133.76 (C-4′), 147.50 (C-1′), 166.09 (C-2), 170.73 (C-5′), 173.23 (C-8′) ppm; IR (KBr): $\bar{\nu}$ = 3352s

(CON*H*), 3050s (Ar–H), 2935w (C–H), 1740s (*CON*), 1631s (C=N) $\rm cm^{-1}$.

N-[4-[[(2-Hydroxyphenyl)methylene]amino]benzoyl]-L-glutamic acid (**3f**)

Yield 40%, mp 165–167°C; MS (EI): m/z=371.53 (molecular weight 370.37); 1 H NMR (300 MHz, DMF-d₇): $\delta=2.41$ (dm, 2H, H-7′), 2.74 (t, 2H, H-8′), 4.87 (m, 1H, H-6′), 7.21 (dd, 2H, H-4, H-5), 7.67 (dd, 1H, H-6), 7.76 (d, 2H, H-2′, J=8.46 Hz), 7.91 (dd, 1H, H-3), 8.31 (d, 2H, H-3′, J=8.46 Hz), 8.90 (d, 1H, CONH), 9.26 (s, 1H, CH=N), 12.88 (b, 1H, Ar–OH), 13.26 (b, 2H, COOH) ppm; 13 C NMR (75 MHz, DMF-d₇): $\delta=26.90$ (C-7′), 30.84 (C-8′), 52.88 (C-6′), 112.93 (C-4), 117.01 (C-5), 119.51 (C-6), 119.73 (C-1), 121.59 (C-2′), 129.46 (C-3′), 133.95 (C-3), 133.29 (C-4′), 151.34 (C-1′), 159.66 (C-2), 165.38 (CH=N), 173.78 (C-5′), 174.33 (C-9′), 174.41 (C-10′) ppm; IR (KBr): $\bar{\nu}=3384$ s (CONH), 3080s (Ar–H), 2933w (C–H), 1734s (CON), 1715s (COO) cm $^{-1}$.

N-[4-[[(2-Hydroxy-4-methoxyphenyl)methylene]amino]benzoyl]-L-glutamic acid (**3g**)

Yield 36%, mp 222–223°C; MS (EI): m/z = 401.26 (molecular weight 400.39); ¹H NMR (300 MHz, DMF-d₇): δ = 2.22 (dm, 2H, H-7′), 2.54 (t, 2H, H-8′), 3.19 (s, 3H, OCH₃), 4.69 (m, 1H, H-6′), 6.56 (d, 1H, H-3), 6.63 (dd, 1H, H-5), 7.53 (d, 2H, H-2′, J = 8.53 Hz), 7.64 (d, 1H, H-6), 8.14 (d, 2H, H-3′, J = 8.53 Hz), 8.72 (d, 1H, CONH), 9.00 (s,1H, CH=N), 12.84 (b, 2H, COOH), 13.58 (b, 1H, Ar-OH) ppm; ¹³C NMR (75 MHz, DMF-d₇): δ = 26.76 (C-7′), 35.66 (C-8′), 52.70 (C-6′), 55.60 (OCH₃), 107.30 (C-5), 100.90 (C-6), 121.34 (C-2′), 129.14 (C-3′), 134.82 (C-3′), 135.52 (C-4′), 132.32 (C-1′), 151.30 (C-1), 152.39 (C-4), 164.02 (C-2), 164.30 (CH=N), 164.75 (C-5′), 166.92 (C-9′), 174.24 (C-10′) ppm; IR (KBr): $\bar{\nu}$ = 3371s (CONH), 3044s (Ar-H), 2947w (C-H), 1708s (CON), 1700s (COO), 1625s (C=N) cm⁻¹.

N-[*4*-(*2*-Thiophenealdimino)benzoy*l*]-*L*-glutamic acid (**3h**) Yield 33%, mp 190–193°C; MS (EI): m/z = 360.74 (molecular weight 360.39); 1 H NMR (300 MHz, *DMF*-d₇): $\delta = 2.34$ (dm, 2H, H-7′), 2.74 (t, 2H, H-8′), 4.83 (m, 1H, H-6′), 7.43 (t, 1H, H-3), 7.54 (d, 2H, H-2′, J = 8.34 Hz), 7.96 (d, 1H, H-2), 8.10 (d, 1H, H-4), 8.37 (d, 1H, H-3′, J = 8.34 Hz), 8.85 (d, 1H, CON*H*), 9.08 (s, 1H, *CH*=N), 12.56 (b, 2H, COO*H*) ppm; 13 C NMR (75 MHz, *DMF*-d₇): $\delta = 26.61$ (C-7′), 30.94 (C-8′), 52.56 (C-6′), 112.90 (C-3), 120.58 (C-2), 121.00 (C-2′), 129.16 (C-3′), 133.30 (C-4′), 134.26 (C-4), 144.19 (C-1′), 154.08 (C-1), 155.12 (*C*H=N), 166.59 (C-5′), 174.03 (C-9′), 174.22 (C-10′) ppm; IR (KBr): $\bar{\nu} = 3307$ s (CON*H*), 3037s (Ar–H), 2935w (C–H), 1718w (*CO*N) cm⁻¹.

N-[*4*-(*2*-*Pyrrolaldimino*)*benzoyl*]-*L*-*glutamic acid* (**3i**) Yield 35%, mp 179°C (decomposed); MS (EI): m/z = 343.47 (molecular weight 343.34); ¹H NMR (300 MHz, *DMF*-d₇): $\delta = 2.03$ (dm, 2H, H-7′), 2.36 (t, 2H, H-8′), 4.40 (m, 1H, H-6′), 6.22 (dd, 1H, H-3), 6.76 (d, 1H, H-2), 7.06 (d, 1H, H-4), 7.25 (d, 2H, H-2′, J = 8.39 Hz), 7.90 (d, 2H, H-3′, J = 8.39 Hz)

8.39 Hz), 8.34 (s, 1H, CH=N), 8.58 (d, 1H, CONH), 11.85 (b, 1H, NH), 12.42 (b, 2H, COOH) ppm; $^{13}\mathrm{C}$ NMR (75 MHz, DMF-d_7): $\delta=26.44$ (C-7'), 30.88 (C-8'), 52.15 (C-6'), 110.99 (C-3), 112.88 (C-2), 120.98 (C-2'), 129.19 (C-3'), 120.91 (C-4'), 129.48 (C-4), 130.95 (C-1'), 152.29 (C-1), 151.86 (CH=N), 166.96 (C-5'), 174.01 (C-9'), 174.43 (C-10') ppm; IR (KBr): $\bar{\nu}=3377\mathrm{s}$ (CONH) , 3048s (Ar–H), 2947w (C–H), 1727s (CON), 1689s (COO), 1628s (C=N) cm $^{-1}$.

N-[4-[[(2-Hydroxy-1-naphtyl)methylene]amino]benzoyl]-L-glutamic acid (**3j**)

Yield 40%, mp 181–183°C; MS (EI): m/z = 420.93 (molecular weight 420.43); ¹H NMR (300 MHz, *DMF*-d₇): $\delta = 2.23$ (dm, 2H, H-7'), 2.57 (t, 2H, H-8'), 4.70 (m, 1H, H-6'), 7.11 (d, 1H, H-4), 7.08 (d, 1H, H-5), 7.44 (t, 1H, H-6), 7.61 (t, 1H, H-7), 7.80 (d, 2H, H-2', $J = 8.60 \,\text{Hz}$), 7.88 (d, 1H, H-8), 8.19 (d, 2H, H-3', $J = 8.60 \,\mathrm{Hz}$), 8.61 (d, 1H, H-3), 8.77 (d, 1H, NH), 9.87 (s, 1H, CH=N), 12.80 (b, 2H, COOH), 15.69 (b, 1H, Ar-OH) ppm; ¹³C NMR (75 MHz, *DMF*- d_7): $\delta = 26.75$ (C-7'), 30.68 (C-8'), 52.71 (C-6'), 109.41 (C-10), 112.92 (C-6), 121.49 (C-2'), 122.16 (C-7), 123.97 (C-8), 124.63 (C-5), 128.50 (C-4), 129.41 (C-3'), 127.58 (C-9), 132.23 (C-4'), 133.77 (C-1), 137.51 (C-3), 147.74 (C-1'), 157.10 (CH=N), 166.46 (C-2), 170.75 (C-5'), 173.78 (C-9'), 174.22 (C-10') ppm; IR (KBr): $\bar{\nu} = 3423$ s (CONH), 3063s (Ar-H), 2947w (C-H), 1721s (CON), 1625s (C=N) cm⁻¹.

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